## Cyclization

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**Gold-Catalyzed Intramolecular Reaction of Indoles with Alkynes: Facile Formation of Eight-Membered Rings and an Unexpected** Allenylation\*\*

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The hydroarylation of alkynes (or alkenylation of arenes) catalyzed by electrophilic transition-metal complexes has emerged as a valuable method for the synthesis of alkenyl arenes and heteroarenes.<sup>[1]</sup> Reetz and Sommer<sup>[2]</sup> as well as Shi and He<sup>[3]</sup> found independently that gold complexes catalyze the intermolecular hydroarylation of alkynes. An intramolecular version was disclosed by Murai, Chatani, and coworkers, who used as catalysts RuII and PtII ions[4] as well as GaCl<sub>3</sub>.<sup>[5,6]</sup> Fürstner et al. reported a similar reaction for the synthesis of phenanthrenes that is catalyzed by PtCl2 and other metal halides.<sup>[7]</sup> Sames and co-workers developed an intramolecular hydroarylation catalyzed by PtCl4 that proceeds under mild conditions.<sup>[8]</sup> Cycloisomerization of ω-aryl-

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1-alkynes has also been performed with  $Hg^{II}$  ions<sup>[9]</sup> or  $Tf_2NH$ .<sup>[10]</sup>

We have recently reported the cyclization of aryl alkynes with Pt<sup>II</sup> or Au<sup>I</sup> catalysts.<sup>[11,12]</sup> Computational studies<sup>[11]</sup> indicate that two pathways compete: a Friedel–Crafts alkenylation and a reaction proceeding through metal cyclopropyl carbenes which show very similar activation energies.

We have now found that, whereas substrates **1** cyclize readily with a cationic gold(t) complex to give azepino[4,5-b]indole derivatives **2**,<sup>[13,14]</sup> the more electrophilic AuCl<sub>3</sub><sup>[15]</sup> leads to indoloazocines **3** by an 8-*endo*-dig process (Scheme 1), a cyclization that has not been observed in

Scheme 1. 7-exo-dig versus 8-endo-dig cyclization of alkynyl indoles 1.

other hydroarylations of alkynes. [16] This type of regiochemical control by the oxidation state of the metal catalyst appears to be unprecedented. [17] Indoloazocine subunit  $\bf 3$  is present in indole alkaloids such as deoxyisoaustamide  $\bf (4)$ , [18,19] okaramine N, [20] and the lundurines (namely, lundurine A  $\bf (5)$ , Scheme 2). [21-23] We also report an unexpected

Scheme 2. Representative indoloazocine alkaloids.

fragmentation reaction that results in the allenylation of the indole nucleus at C2. As predicted by our previous theoretical study,<sup>[11]</sup> indoles tethered to alkynes by two or three atoms undergo 6-*endo*-dig and 6-*exo*-dig cyclizations, respectively.

Several gold complexes and salts, including new Au<sup>I</sup> complexes bearing bulky phosphanes or N-heterocyclic ligands, [24,25] were tested in the intramolecular reaction of indoles with alkynes. In general, the best catalyst for the formation of seven-membered rings **2** is cationic gold(i) complex **7**, [25,26] which allows the cyclizations to be performed

in the absence of Ag<sup>I</sup> salts (Table 1). This complex is readily prepared as an air-stable white solid from the corresponding gold chloride complex. Among the solvents screened (MeNO<sub>2</sub>, acetone, DMF, CH<sub>2</sub>Cl<sub>2</sub>), the best results were

obtained in CH<sub>2</sub>Cl<sub>2</sub>. Thus, reaction of tryptophan derivative 8a with complex 7 as catalyst at room temperature for 30 minutes gave azepino[4,5-b]indole 9a cleanly (Table 1, entry 1). In contrast, the reaction of 8a with AuCl<sub>3</sub> gave indoloazocine 10a cleanly (Table 1, entry 2). Reaction with AuCl also provided 10a, although significant amounts of depropargylated starting material were also obtained (Table 1, entry 3). Reaction of 8a with a catalyst made in situ by chloride abstraction from [AuCl(PPh<sub>3</sub>)] with AgSbF<sub>6</sub> was less selective, and a 1.3:1 mixture of **9a** and 10a was obtained (Table 1, entry 4). Similar results were obtained with 8b and 8c (Table 1, entries 5-9), although in these cases reaction with AuCl<sub>3</sub> gave indoloazocines 10b and 10c along with seven-membered-ring derivatives 11b and 11c, respectively (Table 1, entries 6 and 9). As expected, treatment of 9b with 5 mol % AuCl<sub>3</sub> (CH<sub>2</sub>Cl<sub>2</sub>, room temperature, 16 h) led quantitatively to 11b. N-Allylindole 8d provided seven-membered-ring derivative 9d when 7 was used as the catalyst (Table 1, entry 10). Protic acids do not promote the cyclization of these substrates. Thus, treatment of 8b with para-toluenesulfonic acid (10 mol %) in CH<sub>2</sub>Cl<sub>2</sub> at room temperature for 16 h led only to unchanged starting material.

Substrate **12**, with a tether of only three atoms, reacted satisfactorily with catalyst **7** in CH<sub>2</sub>Cl<sub>2</sub> by a 6-exo-dig pathway to give **13** (Table 2, entry 1), whereas [AuCl(PPh<sub>3</sub>)]/AgSbF<sub>6</sub> gave **13** in lower yield along with dimer **14** (Table 2, entry 2).

The configuration of **14** at the exocyclic double bond was determined by a NOESY experiment. Decomposition was observed when AuCl<sub>3</sub> was used as the catalyst (Table 2, entry 3). Formation of **14** may involve a proton-catalyzed reaction via a tertiary, benzylictype carbocation derived from **12**. Reaction of **15** with an unprotected

propargyl alcohol moiety proceeded uneventfully with Au<sup>I</sup> catalysts to give **16** (Table 2, entries 4 and 5). In contrast, reaction of **15** in the presence of AuCl<sub>3</sub> gave ketone **17**, as a result of isomerization of the exocyclic double bond (Table 2, entry 6).

Amide **18** afforded 5-methylene-4,5-dihydrooxazole **19** under all the conditions examined (Table 2, entries 7–9), although the best results were obtained with catalyst **7**. This type of reactivity has been described recently by Hashmi et al.<sup>[27]</sup> using AuCl<sub>3</sub> as the catalyst. Dihydrooxazole **19** is remarkably stable and does not isomerize to the oxazole under the different reaction conditions. Derivative **20**, a substrate with a tether of only two atoms, reacted with Au<sup>I</sup> catalysts through a 6-*endo*-dig pathway to give **21** (Table 2, entries 10 and 11). In this case, no cyclization was observed with AuCl<sub>3</sub> (Table 2, entry 12).

Surprisingly, when indole **8d** was treated with AuCl<sub>3</sub> (2 mol%) in CH<sub>2</sub>Cl<sub>2</sub> at room temperature for 16 h, allene **22** was obtained as a result of an overall intramolecular allenylation at C-2 of the indole by the *N*-propargyl chain (Scheme 3). Tryptophan derivative **23a** provided indoloazocine **24** and allene **25a** (ca. 1:1 mixture) after being heated at

Table 1: Formation of seven- or eight-membered-ring compounds by cyclization of indoles with alkynes catalyzed by gold complexes.

Entry	Indole	Catalyst <sup>[a]</sup>	t [h]	Product(s)	Yield [%]
1	CO <sub>2</sub> Me N-DNBS 8a	7	0.5	CO <sub>2</sub> Me N DNBS 9a MeO <sub>2</sub> C DNBS	82
2	8 a	AuCl <sub>3</sub>	0.5	DNBS 10a	75
3 4	8a 8a	AuCl [AuCl(PPh <sub>3</sub> )]/AgSbF <sub>6</sub>	1 0.5	10a 9a + 10a (1.3:1) N/SO <sub>2</sub> Ph	70 <sup>[b]</sup> 80
5	N-SO <sub>2</sub> Ph 8b	7	16	9b	65
6	8 b	AuCl <sub>3</sub>	24	10b  + SO <sub>2</sub> Ph  10b  + SO <sub>2</sub> Ph  11b	71
7	8b	[AuCl(PPh <sub>3</sub> )]/AgSbF <sub>6</sub>	16	10b + 11b (5.4:1) 9b + 10b (4:1) ONBS	65
8	N-DNBS 8c	7	16	9c	77
9	8c	AuCl <sub>3</sub>	16	10c + 11b (2.8:1) DNBS 10c  + DNBS 11c	87
10	N-SO <sub>2</sub> Ph 8d	7	0.5	N SO <sub>2</sub> Ph 9d	68

[a] Reactions carried out with 5 mol % of the catalyst in CH2Cl2. [b] N-Depropargylated starting material was also obtained in 23% yield. DNBS = 2,4-dinitrobenzenesulfonyl.

90°C in toluene with catalyst 7. Allene 25b was also obtained as the major compound from 23b.

A rationale for the formation of the allene derivatives is provided in Scheme 4. Eight-membered-ring compound 24 may arise from a 1,2-shift of the initially formed iminium cation I to give II (Scheme 4), [28] which would lose a proton to form III, and then form 24 by protodemetalation. An alternative elimination, facilitated by the electron-withdrawing sulfonyl group R<sup>2</sup>, would yield allenes 25. A similar mechanism is probably followed in the formation of allene 22 from 8d by fragmentation of an AuIII intermediate similar to III. Fragmentation does not occur with the final indoloazocines. Thus, no reaction was observed on treatment of 24 with 7 (5 mol %) in CH<sub>2</sub>Cl<sub>2</sub> at room temperature for 16 h or in toluene at 90°C for 5 h.

Intermediates I and II might also be involved in the formation of indoloazocines 10 a-c (Table 1). A related 1,2-shift could also be involved in the formation of sixand seven-membered-ring compounds by exocyclic pathways, although this seems rather unlikely in the 6-endo-dig cyclization of 20 to form 21 (Table 1, entries 20–21) which would require a 5-endo-dig reaction to form the first spiro intermediate. The different regiochemical outcomes observed in reactions catalyzed by Au<sup>I</sup> complex 7 and AuCl<sub>3</sub> suggests that different mechanisms are involved in these reactions. It is noteworthy that the most electrophilic Au<sup>III</sup> catalyst leads to indoloazocines, which according to PM3 calculations are about 2-5 kcal mol<sup>-1</sup> less stable than their seven-membered-ring isomers 9.

In summary, we have found a facile annulation of six-eightmembered rings on indoles by cyclization with alkynes catalyzed by Au<sup>I</sup> or Au<sup>III</sup> species. Cationic Au<sup>I</sup> complex 7 is the best catalyst for the formation of six- and seven-membered rings through 6endo-dig, 6-exo-dig, and 7-exo-dig cyclizations. Indoloazocines are obtained with AuCl<sub>3</sub> as catalyst through a rare 8-endo-dig process. Surprisingly, allenes are formed by a fragmentation reaction. This allenylation provides a simple

entry into functionalized indole derivatives such as 22 and 25 a-b, which could be used as scaffolds for additional annulation processes.[29]

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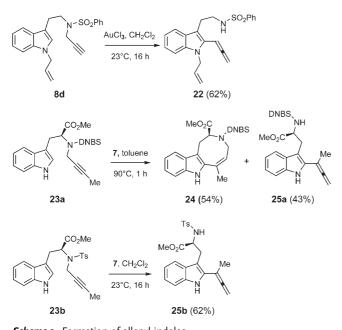
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Table 2: Formation of six-membered-ring compounds by cyclization of indoles with alkynes catalyzed by gold complexes.

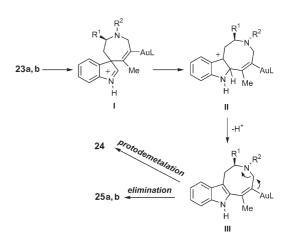
Entry	Indole	Catalyst <sup>[a]</sup>	t [h]	Product(s)	Yield [%]
1	12	7	0.2	13	68
2	∺ 12 12	[AuCl(PPh <sub>3</sub> )]/AgSbF <sub>6</sub>	0.5	н <b>13</b> _[b]	54 <sup>[a]</sup>
3	12	AuCl <sub>3</sub>	2	_[b]	_
4	НО 15	7	0.2	OH 16	72
5	Ме 15	[AuCl(PPh <sub>3</sub> )]/AgSbF <sub>6</sub>	0.2	Me <b>16</b> ♣	60
6	15	$AuCl_3$	0.2	N <sub>Me</sub> 17	100
7	HN 18	7	16	O 19	77
<b>8</b> <sup>[c]</sup>	18	[AuCl(PPh <sub>3</sub> )]/AgSbF <sub>6</sub>	16	19	56
9	18	AuCl <sub>3</sub>	16	19	57
10	20 NMe	7	1	N <sub>Me</sub> 21	92
11	20	[AuCl(PPh <sub>3</sub> )]/AgSbF <sub>6</sub>	16	21	63
12	20	$AuCl_3$	24	_[d]	_

[a] Dimer 14 (25%) was also obtained. [b] Decomposition was observed. [c] Starting material was recovered. [d] Reaction carried out in DMF.



**Scheme 3.** Formation of allenyl indoles.

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**Scheme 4.** Proposed mechanism for the formation of eight-membered-ring compounds **24** and allenes **25**.

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